



Synthesis of Novel Chelating Surfactants for Ion Flotation

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KU Leuven (Leuven, Belgium); start date: 16 Jan 2019

Objectives

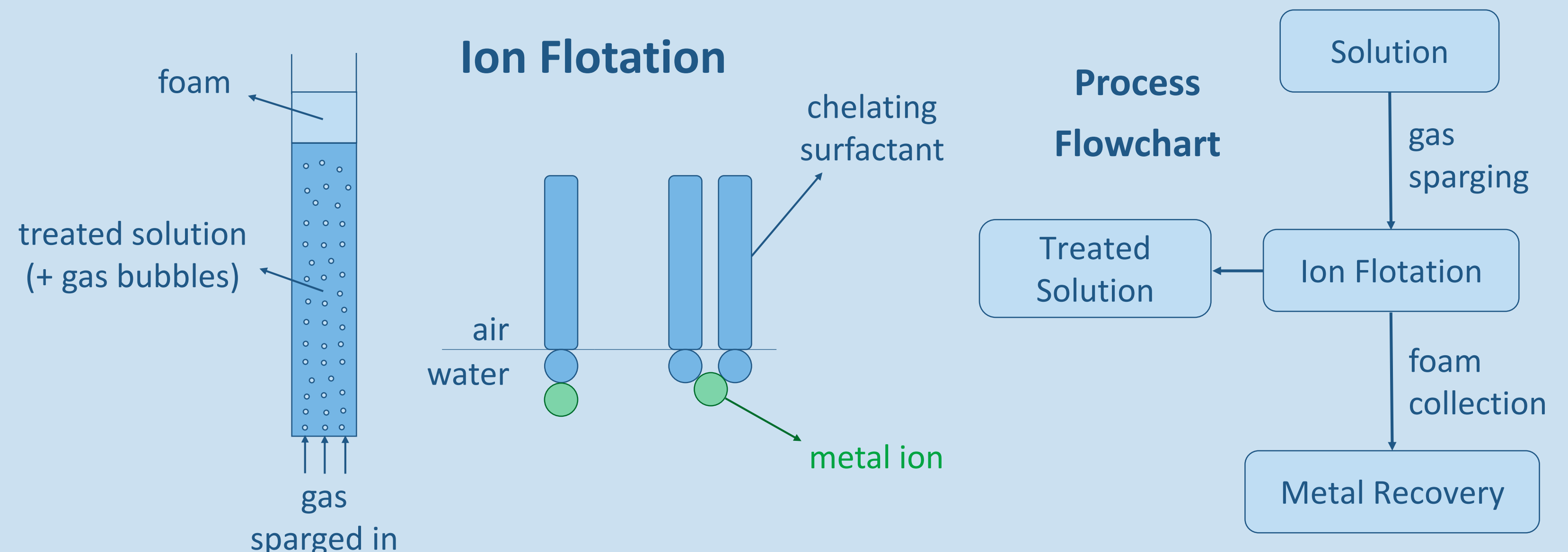
- Design and synthesis of surfactants containing chelating groups using renewable starting materials or abundantly available natural products.
- Elucidation of the efficiency and selectivity of metal removal by ion flotation.
- Study of the recovery of metals from the collected foam.

Milestones

- Aug 2020** – Library of 20 new compounds containing > 4 different ligands including 5 natural product derived surfactants. Initial results on their efficiency in selective ion flotation.
- Dec 2021** – Extended library of compounds and selection of the 5 most promising ones for industrial implementation. Optimized procedure for the selective removal of ions by flotation.

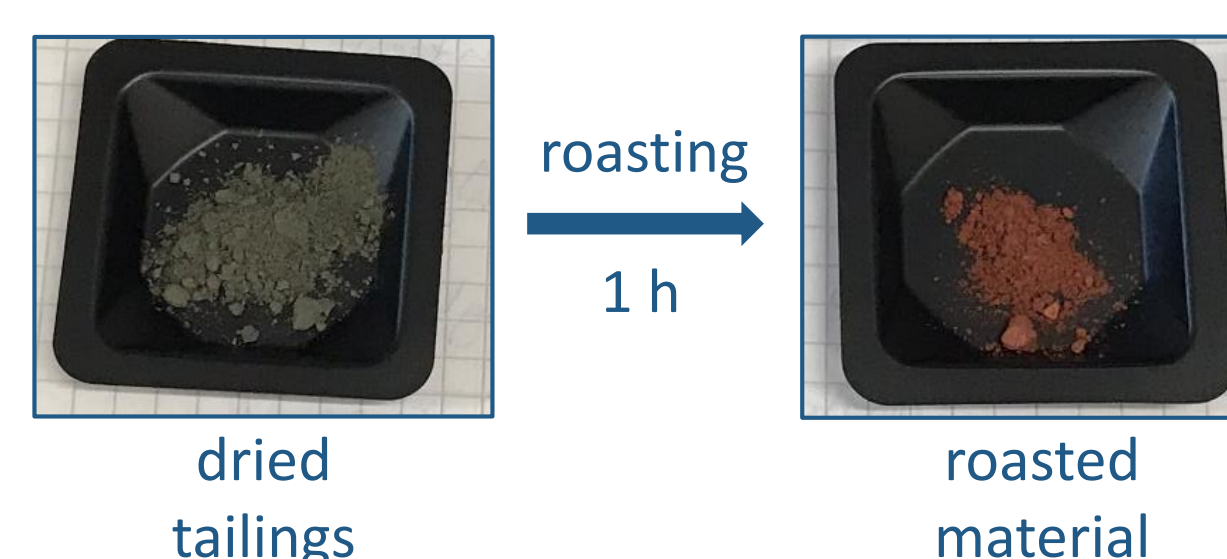
Deliverables Progress

- 1st Scientific paper
- 2nd Scientific paper
- Doctoral thesis



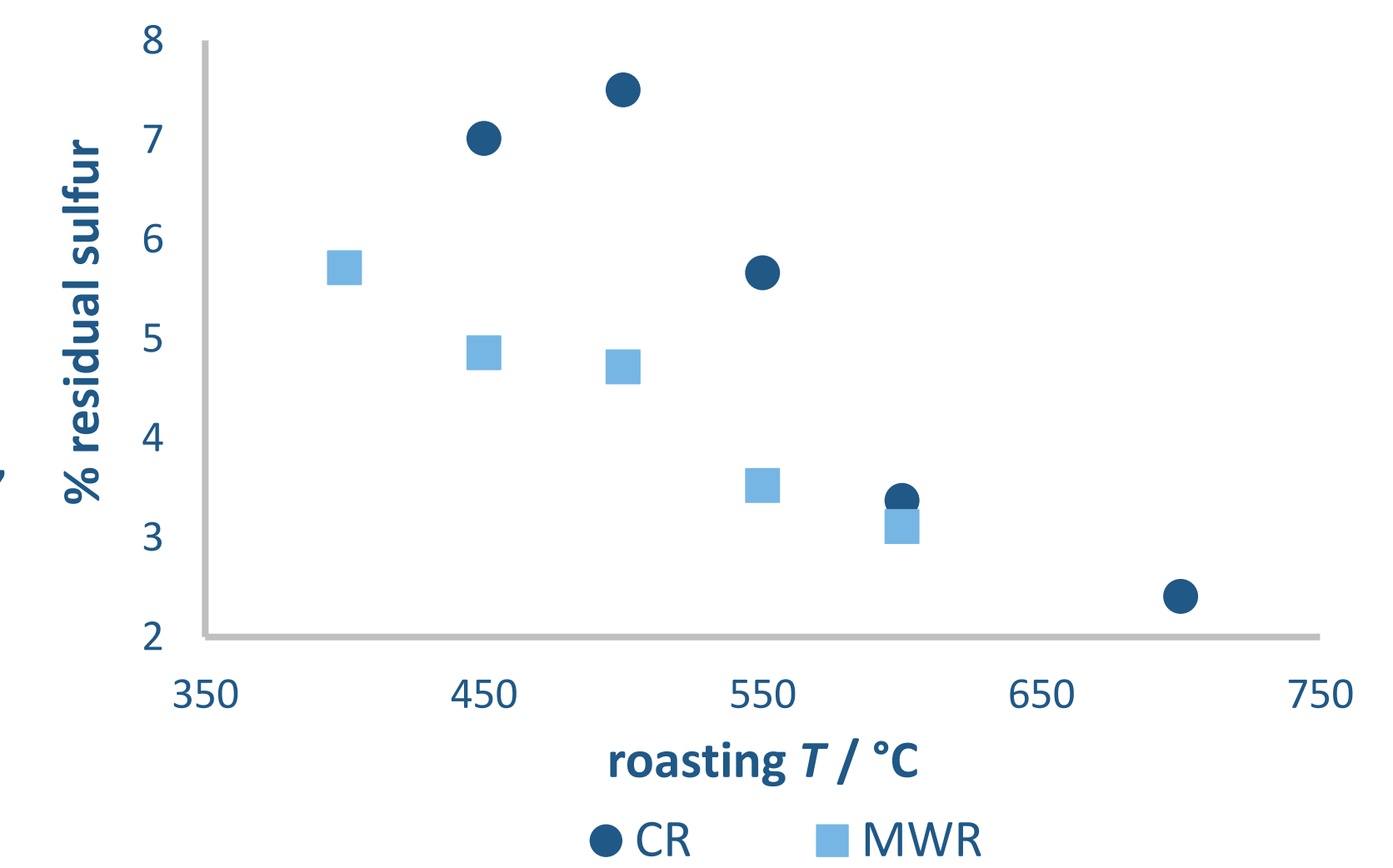
MW Roasting and Leaching of Neves-Corvo Tailings

VITO (Mol, Belgium), supervisor: Jeroen Spooren, 21 Oct – 20 Dec 2019 with Panagiotis Xanthopoulos (ESR10)



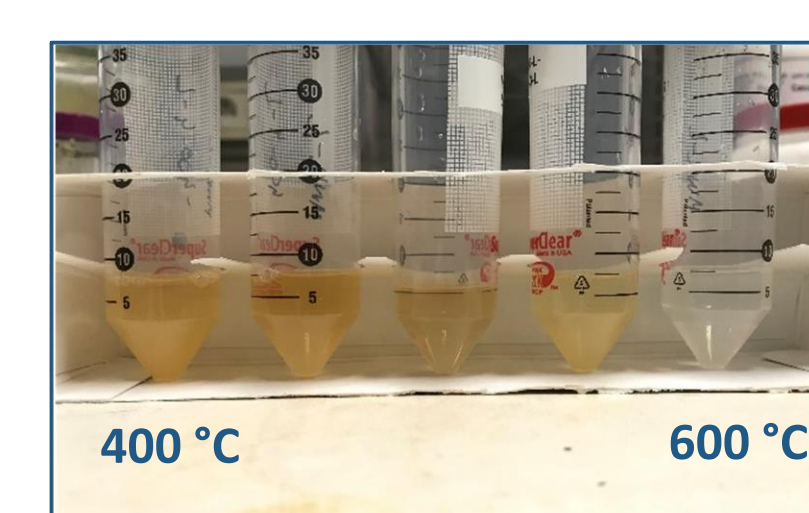
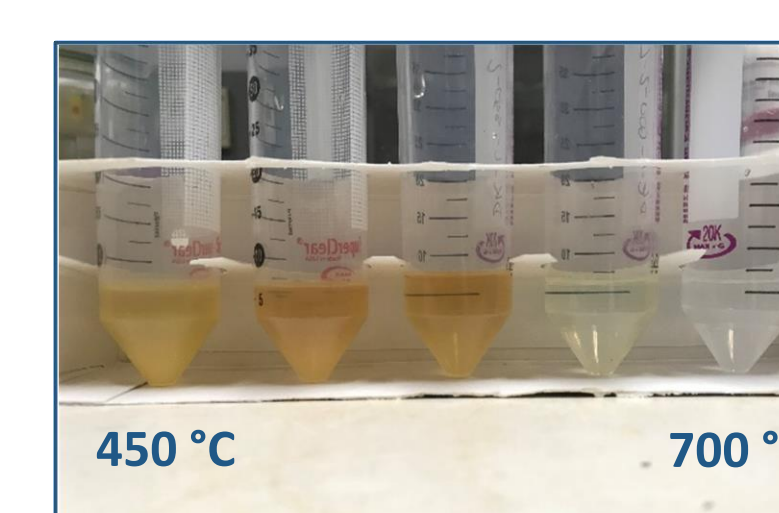
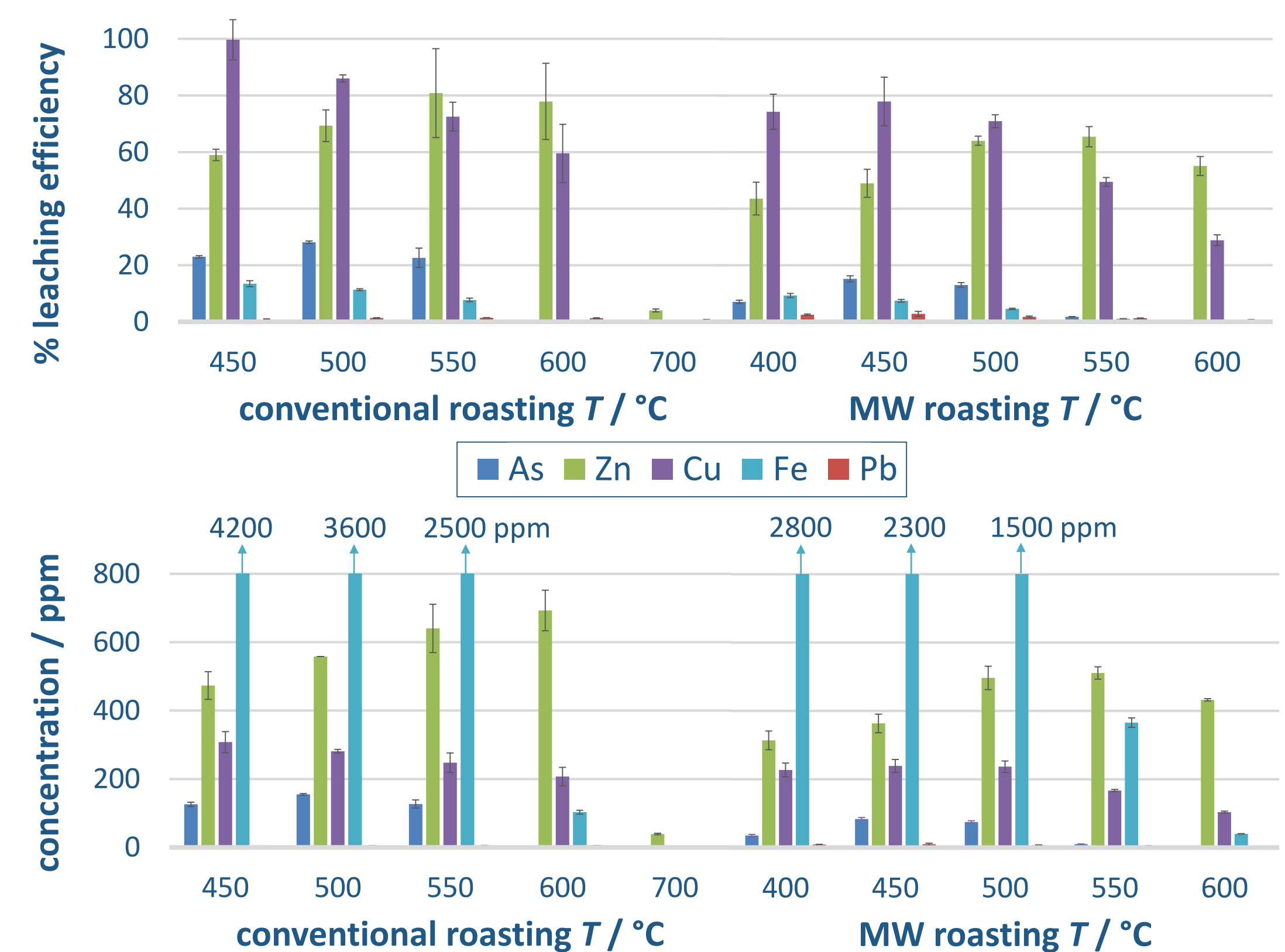
- Sample used: SUL-NC-02 (dried under N₂ at 40 °C), preparation and previous analyses of the starting material were done by Maja Vučković (ex ESR7).
- Conditions: $T = 400 - 700$ °C, $t = 60$ min, $m(\text{CR}) = 5.0$ g in porcelain cup crucibles, $m(\text{MWR}) = 6.0$ g in alumina boat crucibles, samples were ground prior to roasting.

- Characterization: Starting material – XRD, XRF, TGA; Roasted material – XRD, XRF; Leachates – ICP-OES; Solid residues – XRF
- Crucial parameters: roasting t and T , size and shape of the crucible, position within the oven.



Water Leaching

Conditions: L/S = 10, $t = 30$ min, $T = \text{r.t.}$



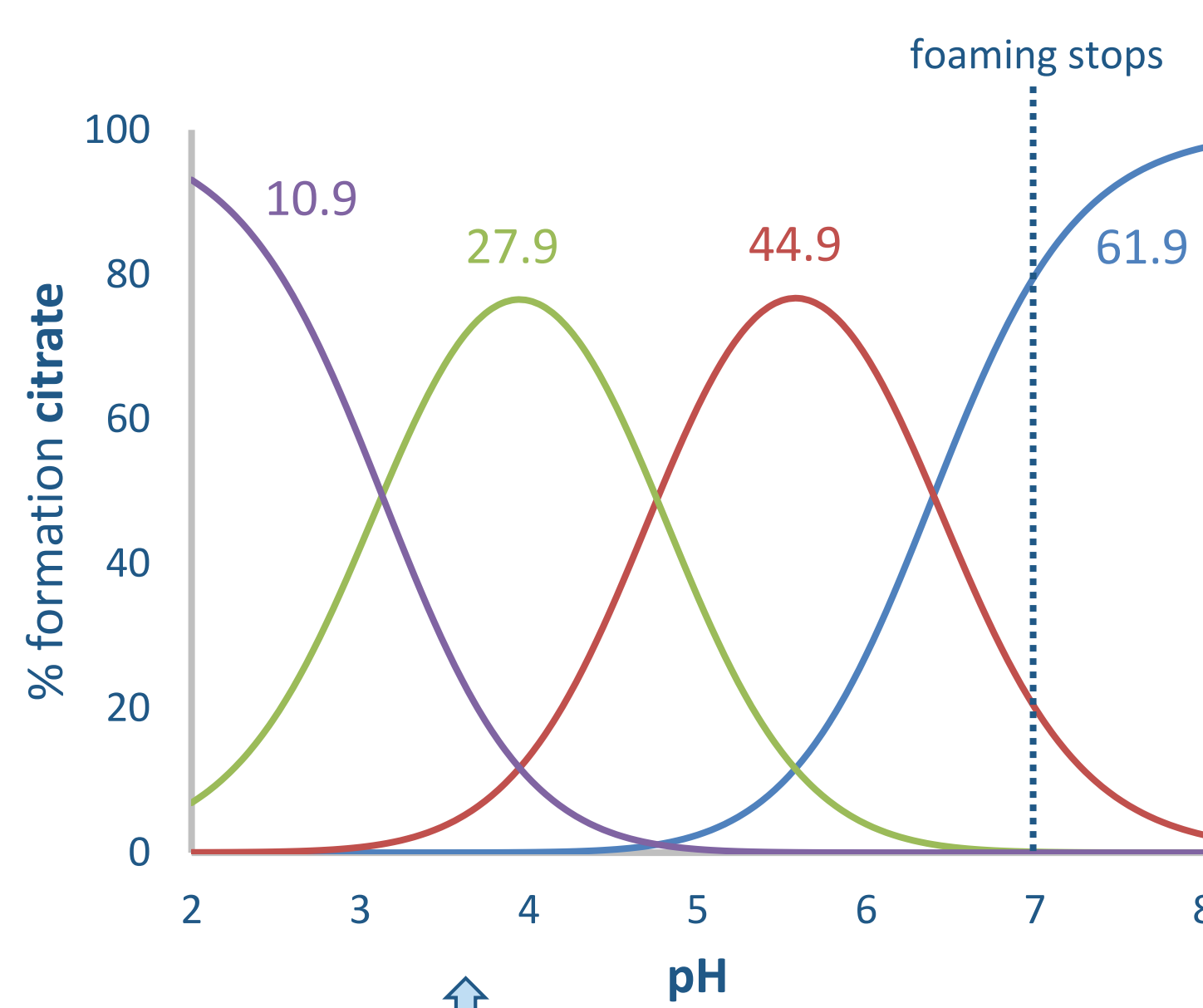
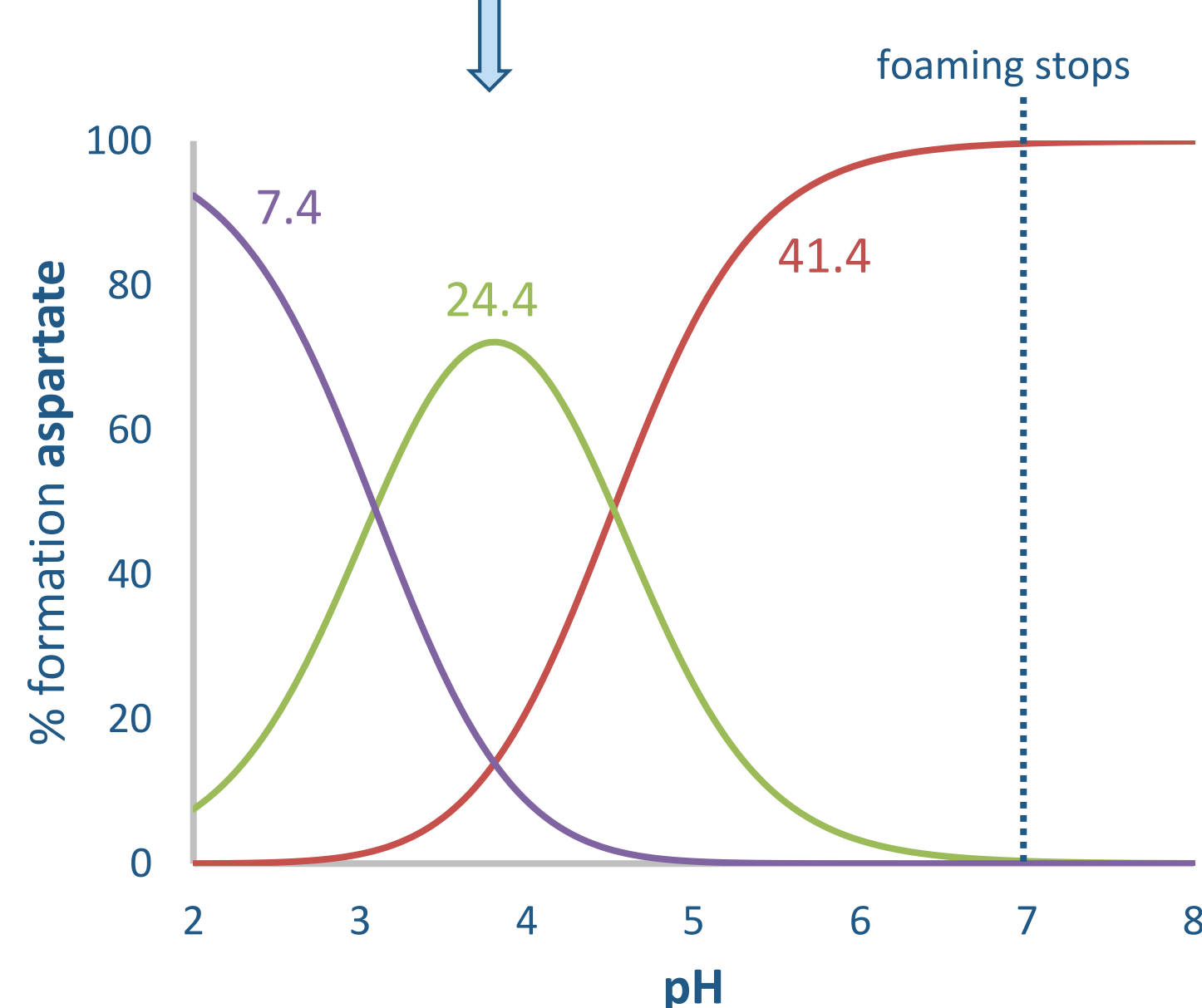
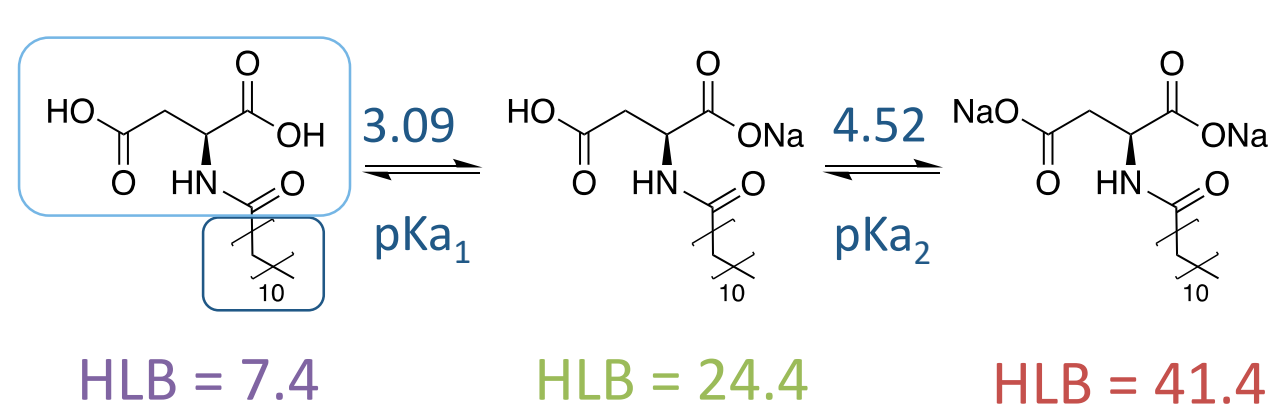
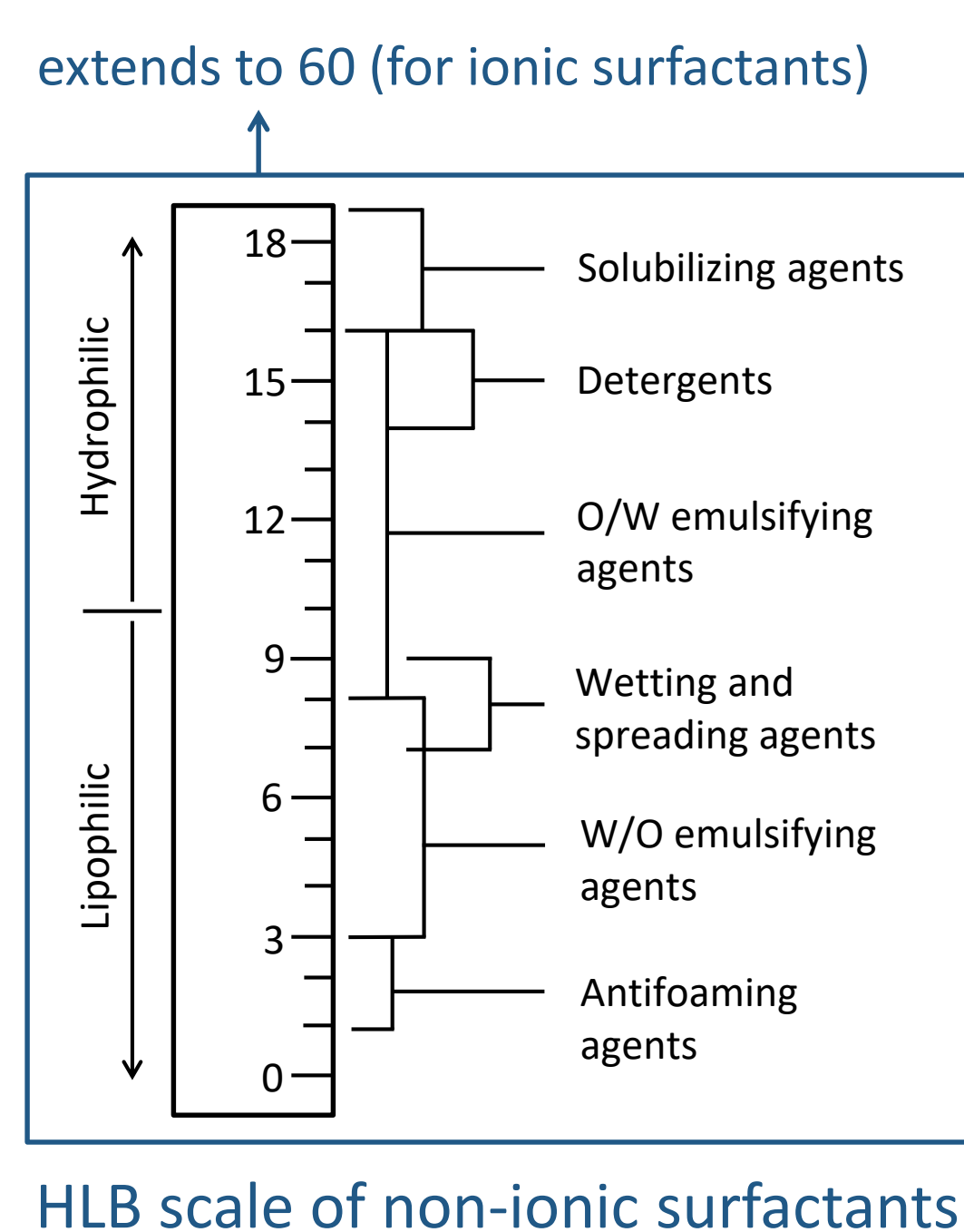
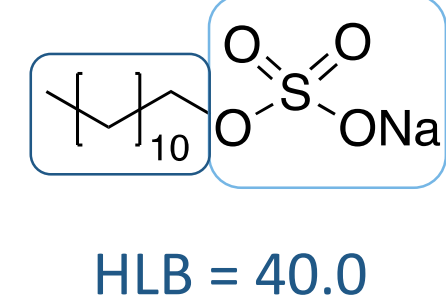
Design of Chelating Surfactants - HLB approach

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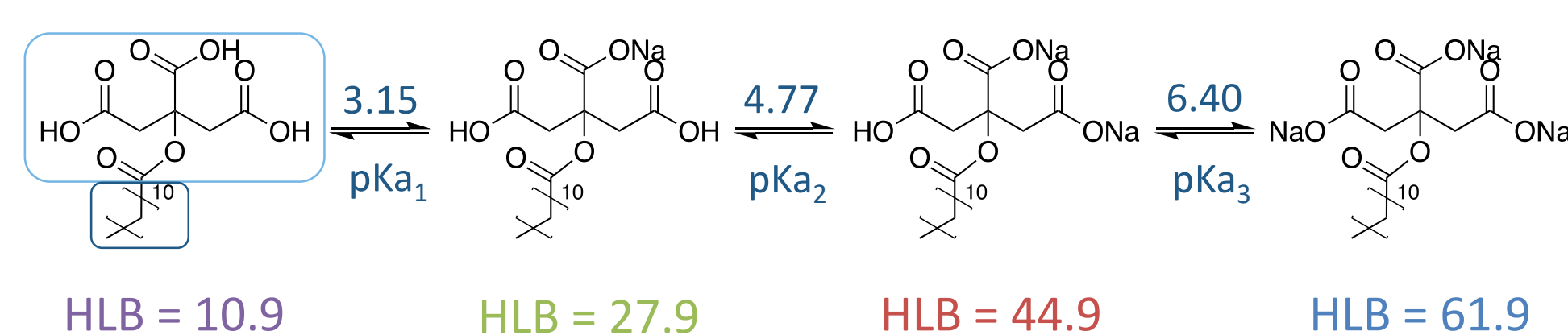
- Modelling of the structure to surface activity relation by hydrophilic-lipophilic balance (HLB) – an empirical parameter.

$$\text{HLB (Davies)} = n_H \times X_H - n_L \times Y_L + 7$$

n_H, n_L – respective number of hydrophilic and lipophilic groups
 X_H, Y_L – empirical numbers assigned to each hydrophilic and lipophilic group, respectively



HLB range of foam-forming species: **21 – 28**



Conclusion

- HLB model describes the structure to surface activity relation well, but there are other structural characteristics influencing the foaming properties such as vicinity of the charged groups within the molecule.
- The model might be improved using computational methods.
- HLB is affected by the chelation of metal ions – foaming properties need to be investigated in aqueous solutions containing metal ions.

Upcoming Steps

- Ion Flotation: single element solution experiments – metal removal efficiency and foaming properties in the presence of metal ions.
- Surfactant Characterization – CMC determination at 3 distinct pH values to gain more insight into the influence of pH.
- Synthesis of histidine and glutathione surfactants.

